

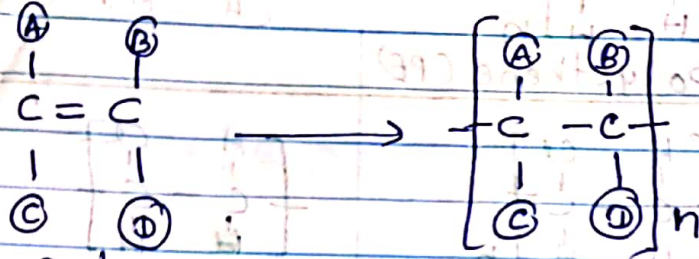
සාම ප්‍රවේශයක් ලෙස හා ම විදිසුන් අලුත් විදුලිකීන් ආරම්භ කරන්න. ඉස්ලාමාදා විකාශයක් ලෙස විකාශයක් ලෙස ප්‍රිය පසුකරන්න. ආදිය.

ප්‍රශ්න පොතේ විකාශයක්

විකාශ පොතේ විකාශයක්

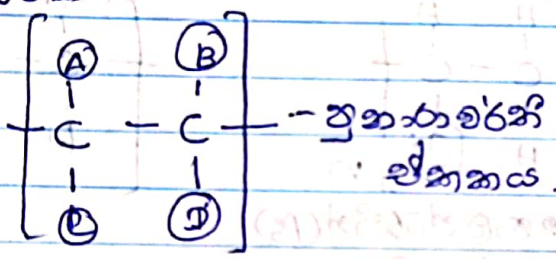
බහුඅවයවික

ඒකාමයවික ලෙස හැඳින්වෙන සාපේක්ෂව කුඩා රසායනික අණු විශාල සංඛ්‍යාවක් ඒකමාන රසායනික බැදීමෙන් අනිවාර්යවෙන්ම බහුඅවයවික වේ.



තැනුම් ඒකකය
ඒකාමයවික

බහුඅවයවික



බහුඅවයවීකරණයේදී සිදුවන ප්‍රතික්‍රියාව යනු ඒකකය මත බහුඅවයවික ආකාර 02 ක්.

1. ආකලන බහුඅවයවික
2. සංගණන බහුඅවයවික

1. ආකලන බහුඅවයවික

ප්‍රශ්න අංකය:

විෂය අංකය:

අනුමාපන අංකය	ඒකාණුක ඒකකය	බහුඅණුක ඒකකය	ඉන්ද්‍රජාලීන ඒකකය
5	$\begin{array}{c} \text{H} & \text{H} \\ & \\ \text{C} = & \text{C} \\ & \\ \text{H} & \text{H} \end{array}$ <p>ethene</p>	$\left[\begin{array}{c} \text{H} & \text{H} \\ & \\ -\text{C} - & \text{C}- \\ & \\ \text{H} & \text{H} \end{array} \right]_n$ <p>polyethene (PE)</p>	$\left[\begin{array}{c} \text{H} & \text{H} \\ & \\ \text{C} - & \text{C} - \\ & \\ \text{H} & \text{H} \end{array} \right]$
10	$\begin{array}{c} \text{H} & \text{Cl} \\ & \\ \text{C} = & \text{C} \\ & \\ \text{H} & \text{H} \end{array}$ <p>ඔස්නයිල අලොරයිඩ්</p>	$\left[\begin{array}{c} \text{H} & \text{Cl} \\ & \\ -\text{C} - & \text{C}- \\ & \\ \text{H} & \text{H} \end{array} \right]_n$ <p>PVC</p>	$\left[\begin{array}{c} \text{H} & \text{Cl} \\ & \\ \text{C} - & \text{C} - \\ & \\ \text{H} & \text{H} \end{array} \right]$
15	$\begin{array}{c} \text{H} & \text{H} \\ & \\ \text{C} = & \text{C} \\ & \\ \text{H} & \text{C}_6\text{H}_5 \end{array}$ <p>ස්ටයරීන්</p>	$\left[\begin{array}{c} \text{H} & \text{H} \\ & \\ -\text{C} - & \text{C}- \\ & \\ \text{H} & \text{C}_6\text{H}_5 \end{array} \right]_n$ <p>නොලිස්ටරීන් (PS) කාජුතොලී.</p>	$\left[\begin{array}{c} \text{H} & \text{H} \\ & \\ \text{C} - & \text{C} - \\ & \\ \text{H} & \text{C}_6\text{H}_5 \end{array} \right]$
20	$\begin{array}{c} \text{F} & \text{F} \\ & \\ \text{C} = & \text{C} \\ & \\ \text{F} & \text{F} \end{array}$ <p>ටෙෆ්ලන් ප්ලෝස්ටික්</p>	$\left[\begin{array}{c} \text{F} & \text{F} \\ & \\ -\text{C} - & \text{C}- \\ & \\ \text{F} & \text{F} \end{array} \right]_n$ <p>නොලිස්ටරීන් ප්ලෝස්ටික් (PTFE) ටෙෆ්ලන්</p>	$\left[\begin{array}{c} \text{F} & \text{F} \\ & \\ \text{C} - & \text{C} - \\ & \\ \text{F} & \text{F} \end{array} \right]$

- ආකලන බහුඅණුක ඒකකවල දිසුනරණ -

25

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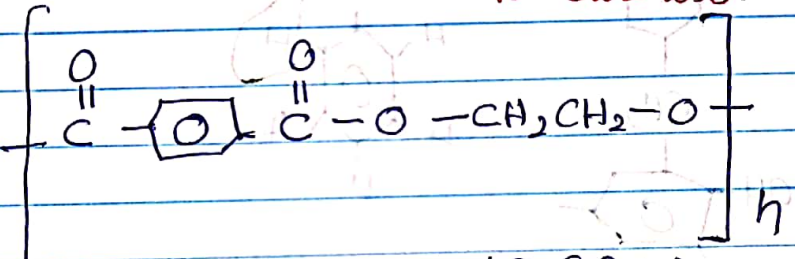
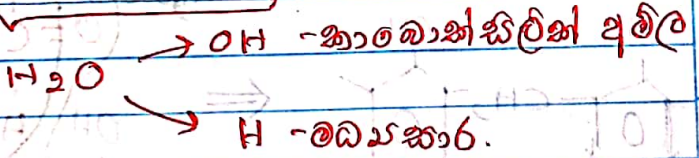
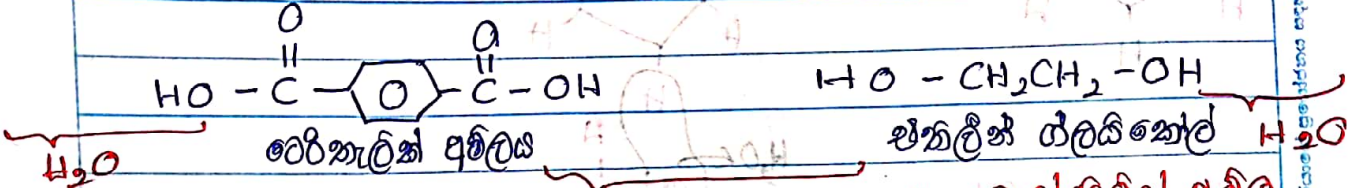
ප්‍රශ්න අංකය:

විෂය අංකය:

2. සංගතන බහුඅවයවික.

සංගතන ප්‍රතික්‍රියාවක් ඒකකරණ (පිප්ටර් සෑදීම හෝ ඇමයිඩ් සෑදීම) තුළ පුනුක සන්නිධයක් සහිත පුනු මුණක් කරමින් ඇතිවන බහුඅවයවික සංගතන බහුඅවයවික වේ.

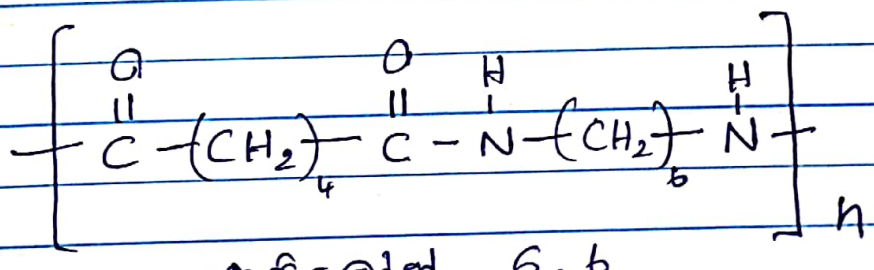
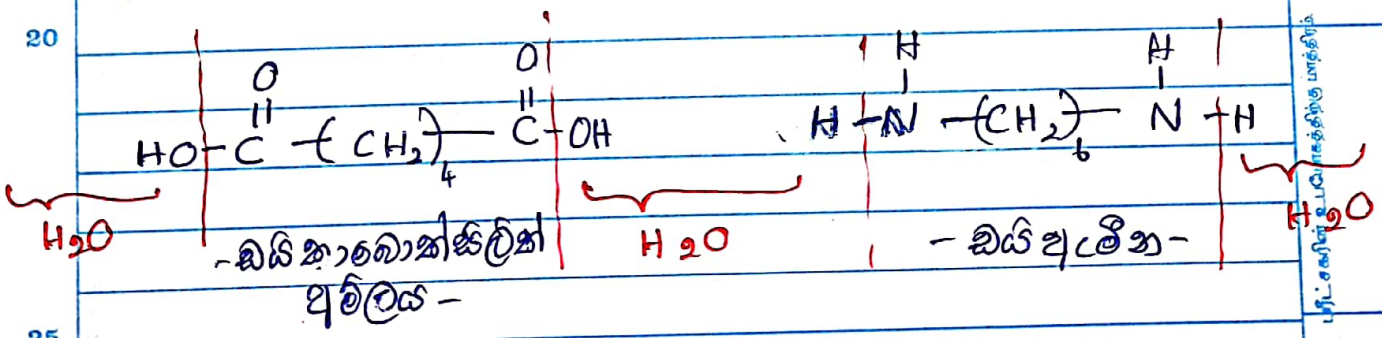
උදා:- 1. තොලිපිප්ටර් :- eg :- තොලිපිප්ටරික් ටෙරිනාලේට්.



තොලිපිප්ටරික් ටෙරිනාලේට් (PET)

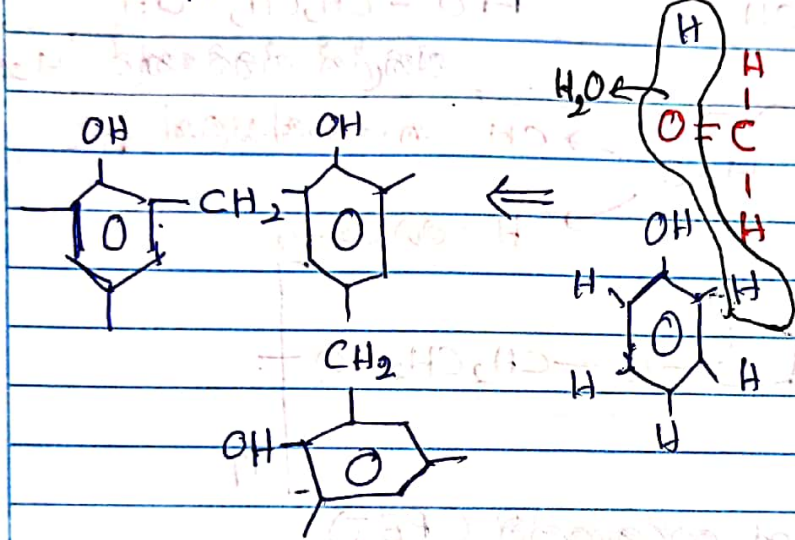
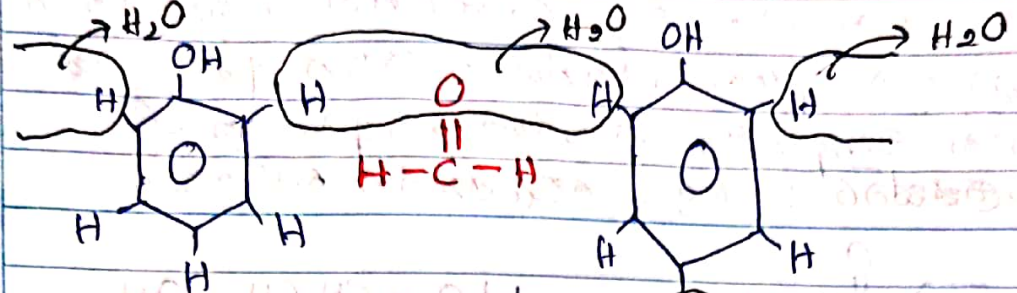
2. තොලිපිප්ටරික්

උදා:- නයිලෝන් 6,6



නයිලෝන් 6,6

3. രീതികൾ (രീതികൾ - തിരിച്ചറിയുന്നതിന്)



രണ്ടു കൃത്യമായ രീതികൾ ജന്മദാർശി രീതികൾ ഉണ്ടാക്കി തിരിച്ചറിയുന്നതിന് ഉപയോഗിക്കാവുന്ന വിധം ചുരുക്കി എഴുതാം.

പിന്നം നമ്പർ:

പിന്നം നമ്പർ:

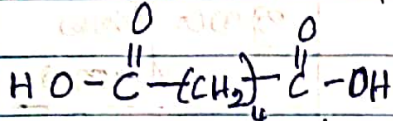
പൊതുവിവരണം

പദപദപദപദപദ	പൊതുവിവരണം	പുനഃനിർമ്മാണ രീതി
<p>5</p> <p>(01)</p> $\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{C} = \text{C} \\ \quad \\ \text{H} \quad \text{H} \end{array}$ <p>- ethene -</p>	$\left[\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ - \text{C} - \text{C} - \\ \quad \\ \text{H} \quad \text{H} \end{array} \right]_n$ <p>- polyethene - PE</p>	$\left[\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ - \text{C} - \text{C} - \\ \quad \\ \text{H} \quad \text{H} \end{array} \right]$
<p>10</p> <p>(02)</p> $\begin{array}{c} \text{H} \quad \text{Cl} \\ \quad \\ \text{C} = \text{C} \\ \quad \\ \text{H} \quad \text{H} \end{array}$ <p>- മീഥേൻ ക്ലോറൈഡ്</p>	$\left[\begin{array}{c} \text{H} \quad \text{Cl} \\ \quad \\ - \text{C} - \text{C} - \\ \quad \\ \text{H} \quad \text{H} \end{array} \right]_n$ <p>PVC</p>	$\left[\begin{array}{c} \text{H} \quad \text{Cl} \\ \quad \\ - \text{C} - \text{C} - \\ \quad \\ \text{H} \quad \text{H} \end{array} \right]$
<p>15</p> <p>(03)</p> $\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{C} = \text{C} \\ \quad \\ \text{H} \quad \text{C}_6\text{H}_5 \end{array}$ <p>- ബെൻസീൻ</p>	$\left[\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ - \text{C} - \text{C} - \\ \quad \\ \text{H} \quad \text{C}_6\text{H}_5 \end{array} \right]_n$ <p>- ബെൻസീൻ പൊളിമർ</p>	$\left[\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ - \text{C} - \text{C} - \\ \quad \\ \text{H} \quad \text{C}_6\text{H}_5 \end{array} \right]$
<p>20</p> <p>(04)</p> $\begin{array}{c} \text{F} \quad \text{F} \\ \quad \\ \text{C} = \text{C} \\ \quad \\ \text{F} \quad \text{F} \end{array}$ <p>- ടെറ്റ്രാഫ്ലൂറൈഡ്</p>	$\left[\begin{array}{c} \text{F} \quad \text{F} \\ \quad \\ - \text{C} - \text{C} - \\ \quad \\ \text{F} \quad \text{F} \end{array} \right]_n$ <p>- ടെറ്റ്രാഫ്ലൂറൈഡ് പൊളിമർ</p>	$\left[\begin{array}{c} \text{F} \quad \text{F} \\ \quad \\ - \text{C} - \text{C} - \\ \quad \\ \text{F} \quad \text{F} \end{array} \right]$
<p>25</p> <p>(05) ബെൻസീൻ</p> $\text{HO} - \overset{\text{O}}{\parallel}{\text{C}} - \text{C}_6\text{H}_4 - \overset{\text{O}}{\parallel}{\text{C}} - \text{OH}$ <p>- ബെൻസീൻ ടെറ്റ്രാകാർബോക്സിലിക്</p>	$\left[\overset{\text{O}}{\parallel}{\text{C}} - \text{C}_6\text{H}_4 - \overset{\text{O}}{\parallel}{\text{C}} - \text{O} - \text{CH}_2\text{CH}_2 - \text{O} \right]_n$ <p>- ബെൻസീൻ ടെറ്റ്രാകാർബോക്സിലിക് പൊളിമർ</p>	$\left[\overset{\text{O}}{\parallel}{\text{C}} - \text{C}_6\text{H}_4 - \overset{\text{O}}{\parallel}{\text{C}} - \text{O} - \text{CH}_2\text{CH}_2 - \text{O} \right]$
<p>30</p> <p>(06) ടെറ്റ്രാഹൈഡ്രോ</p> $\text{HO} - \text{CH}_2\text{CH}_2 - \text{OH}$ <p>- ടെറ്റ്രാഹൈഡ്രോ</p>	<p>- ടെറ്റ്രാഹൈഡ്രോ</p> <p>ബെൻസീൻ - PET</p>	

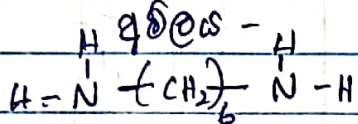
ප්‍රශ්න අංකය:

විභාග අංකය:

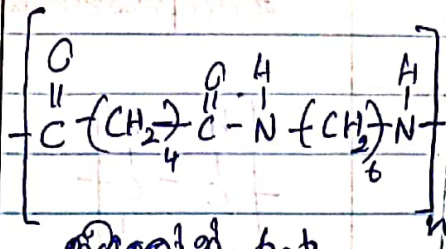
(06) නොලී ඇමෝනියා :



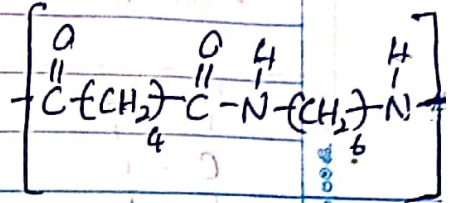
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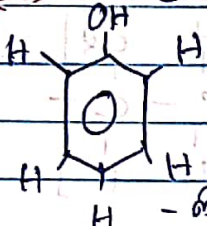
- ඔක්සලීන් -



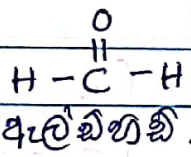
නියමයන් 6,6



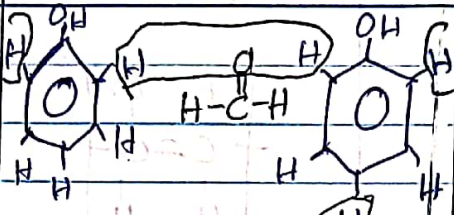
(07) හීන් ලයට්.



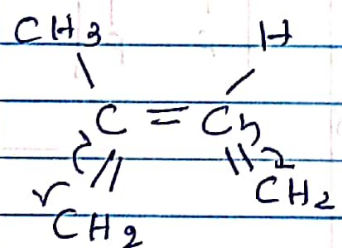
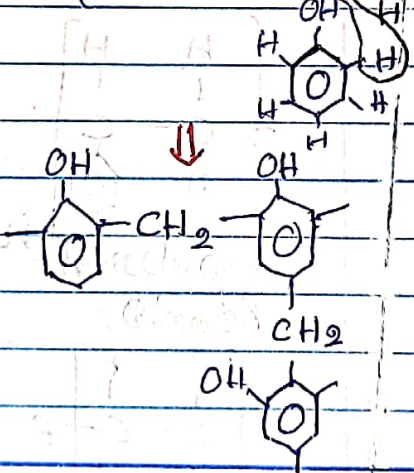
- හීන් ලය -



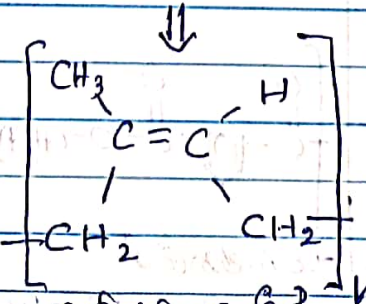
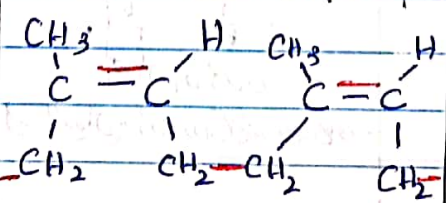
- ඇලඩිහයිඩ්.



- හීන් ලය
හේමිල්ඩ් හයිඩ් -



- ඇයිනොලීන් -



නොලී ඇයිනොලීන්

